The antiferromagnetic insulator Ca₃FeRhO₆: characterization and electronic structure calculations

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We investigate the antiferromagnetic insulating nature of $\text{Ca}_3\text{FeRhO}_6$ both experimentally and theoretically. Susceptibility measurements reveal a Néel temperature $T_N \simeq 20\,\text{K}$, and a magnetic moment of $5.3\mu_B/\text{f}$. u., while Mössbauer spectroscopy strongly suggests that the Fe ions, located in trigonal prismatic sites, are in a 3+ high spin state. Transport measurements display a simple Arrhenius law, with an activation energy of $\sim 0.2\,\text{eV}$. The experimental results are interpreted with LSDA band structure calculations, which confirm the Fe³⁺ state, the high-spin/low-spin scenario, the antiferromagnetic ordering, and the value for the activation energy.

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I. INTRODUCTION

Interest in transition metal oxides has never been restricted to the most spectacular phenomenon of the high- T_c superconductivity, but also concerns, inter alia, metal-insulator transitions, colossal magnetoresistance, and magnetic and orbital ordering [1]. Among the numerous magnetic transitions that have been studied, most of them are associated with structural transitions In contrast, no anomalous temperature dependence of the structural parameters has been reported in the m = 0, n = 1 members of the oxides family $A_{3n+3m}A'B_{3m+n}O_{9m+6n}$, which are currently attracting much attention [3]. These A₃A'BO₆ compounds crystallize in the K₄CdCl₆ structure which consists of infinite chains along the c-axis made of a 1:1 alternation of face-shared trigonal prisms $(A'O_6)_{TP}$ and octahedra (BO₆)_{oct}. According to their rhombohedral symmetry, the A cations separate the chains, the latter forming a hexagonal array. For such compounds, when the A' trigonal prism (TP) site is occupied by a magnetic cation, this provides interesting physical properties created by the coexistence of one-dimensionality character and geometrical frustration. This is illustrated by Ca₃Co₂O₆ for which the ordered antiferromagnetic state below $T_N \sim 26\,\mathrm{K}$ [4, 5] bears some similarity to the partially disordered antiferromagnetic (PDA) state as originally proposed for ABX₃ geometrically frustrated 1D compounds [6]. But in marked contrast with the ABX₃ members, the intrachain coupling in Ca₃Co₂O₆ is ferromagnetic [4] and the magnetic field induced magnetization is very spectacular [7, 8, 9, 10]. Indeed, as a function of the applied magnetic field, several magnetization jumps with a constant

field spacing are observed. Besides, the saturation magnetization is larger than expected from the assumption of different spin states for Co³⁺ high spin (HS) and low spin (LS) in the TP and oct., respectively. Nonetheless, the ferromagnetic coupling along the chains is probably related to this "spin state ordering", the latter resulting from the different crystalline electrical fields in each Co³⁺ polyhedron. Such a coupling is likely to involve both LS Co^{3+} and O ions [11, 12, 13]. In that respect, the different magnetic behavior of the two isostructural compounds Ca₃FeRhO₆ and Ca₃CoRhO₆ is worth mentioning [14, 15, 16, 17, 18, 19, 20, 21]. In the latter, the ferromagnetic intrachain coupling is expected as ${\rm Rh}^{3+}$ is isoelectronic to Co³⁺ (3d6), whereas the global magnetic behavior of Ca₃FeRhO₆ appears to be antiferromagnetic although Fe³⁺ (d5) or Fe²⁺ (d6) are both HS cations with large S values (5/2 or 2). Such different background states for the Ca_3MRhO_6 1D compounds (with M = Feand M = Co) suggest subtle changes of the electronic structure.

In Ca₃FeRhO₆, contradicting results have been reported for the oxidation states of iron and rhodium cations which add more complexity to the interpretation. In order to shed light on the magnetic and electronic behavior of Ca₃FeRhO₆, we compare, in continuation of previous work [22], results of band structure calculations to the electrical and magnetic properties, together with Mössbauer spectroscopy measurements.

II. EXPERIMENTS

The polycrystalline sample of Ca₃FeRhO₆ was prepared by mixing the precursors CaO, Fe₂O₃ and Rh₂O₃ in the molar ratios 3:0.5:0.5. The thoroughly mixed powder, pressed in bars ($\sim 2 \times 2 \times 10 \text{mm}^3$) was first heated at 900°C for 24h and then at 1250°C for a $3 \times 24 \,\mathrm{h}$ period with intermediate X-ray controls. The crystallinity and purity of the obtained black product were checked by X-ray powder diffraction. The diffraction peaks have been indexed in the space group R-3c with a and c values very close to those reported in Refs. [14, 15], allowing to refine the 3:1:1 ratio for the cation Ca: Fe: Rh with an uncertainty of approximately 3 % which is acceptable in all respects. Besides, small intensity peaks were also found, that could be attributed to Ca₂Fe₂O₅. Magnetic measurements were performed with a SQUID magnetometer. Electrical resistivity was measured by the four probe technique. The four electrical contacts were ultrasonically deposited on a bar. The measurements were made by using a physical properties measurement system (PPMS). The ⁵⁷Fe powder Mössbauer resonance spectrum at room temperature was performed with a transmission geometry by use of a constant acceleration spectrometer and a γ -ray source from ⁵⁷Co embedded in a rhodium matrix. The velocity scale was calibrated with an α -Fe foil at room temperature. The spectra were fitted with Lorentzian lines by the unpublished MOSFIT program. The isomer shift was referred to metallic α -Fe at 293 K.

III. RESULTS

A. Magnetism

When compared to the T-dependent reciprocal magnetic susceptibility curve $[\chi^{-1}(T)]$ of Ca₃Co₂O₆ as given in Ref. [9], the $\chi^{-1}(T)$ curve of Ca₃FeRhO₆ (Fig. 1) exhibits a much more linear behavior extending over a larger T range. This result reflects a lack of ferromagnetic interactions for the latter as also attested by the different extrapolated temperatures for $\chi^{-1}(T=\theta_{\rm CW})=0$, the Curie-Weiss temperature $\theta_{\rm CW}$ values being $-20\,{\rm K}$ and +25 K for Ca₃FeRhO₆ and Ca₃Co₂O₆, respectively. Furthermore in Ca₃Co₂O₆, a χ^{-1} drop below $T_N \sim 26$ K is observed, which indicates that a net ferrimagnetic state is reached: on each triangle made by three neighboring CoO₆ chains, two chains are antiferromagnetically coupled (zero net magnetic moment) whereas the third one exhibits a net ferromagnetic magnetization along the direction of the external applied magnetic field. In contrast, the $\chi^{-1}(T)$ curve of Ca₃FeRhO₆ exhibits a χ^{-1} increase below $\sim 20\,\mathrm{K}$ indicative of a 3D antiferromagnetic phase. Analysis of the slope along the linear region yields an effective paramagnetic moment $\mu_{\text{eff}}(\exp)$ = $5.3\mu_B/(Fe+Rh)$. We will refer to this experimental value below in subsection C.

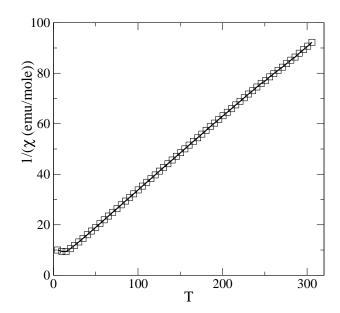


FIG. 1: Susceptibility of Ca₃FeRhO₆.

The RT Mössbauer spectrum of this compound consists of a paramagnetic doublet and is consistent with the measurements reported in Ref. [16]. However, the best fit was obtained with two Mössbauer components, the hyperfine parameters of which are given in Table I. The observed isomer shift value ($IS = 0.45 \pm 0.1 \,\mathrm{mm/s}$) of the main component ($\% = 96 \pm 2$) is typical of Fe³⁺ ions. Its high absolute quadrupole splitting value ($QS = 1.20 \,\mathrm{mm/s}$) shows that this site is not in the octahedral symmetry, and therefore the Fe ions are located in trigonal sites. For the minor Mössbauer component ($\% = 4 \pm 2$), the IS value of $0.73 \pm 0.1 \,\mathrm{mm/s}$ is rather corresponding to Fe²⁺ ions. Therefore the vast majority of iron ions in Ca₃FeRhO₆ are in the trivalent state.

B. Transport

The second set of measurements concerns the expected localized nature of the electrical transport. Indeed, as shown in Table II, Ca₃FeRhO₆ appears far more insulating than the related Co compounds since, for instance at 300 K, the resistivity ρ for Ca₃FeRhO₆ is 160 times larger than that of Ca₃Co₂O₆ [23]. The T-dependence of the resistivity confirms that Ca₃FeRhO₆ is insulating, as

TABLE I: Refined ⁵⁷Fe Mössbauer hyperfine parameters of Ca_3FeRhO_6 at room temperature including the linewidth Σ , and %: relative intensity of the Mössbauer site.

IS (mm/s)	$\Sigma (\mathrm{mm/s})$	QS (mm/s)	%
0.45(1)	0.28(1)	1.20(1)	96 (2)
0.73 (1)	0.36(1)	1.49 (1)	4 (2)

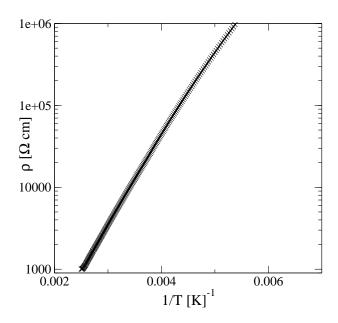


FIG. 2: Temperature dependence of the resistivity of Ca_3FeRhO_6 , yielding an activation temperature $T_0 = 2200K$.

shown in Fig. 2. As T decreases, ρ increases very rapidly in Ca₃FeRhO₆ reaching the set-up limit (corresponding to $\sim 10^6\Omega$) at $\sim 230\,\mathrm{K}$. For the available T-range, the linear $\ln\rho$ (T^{-1}) curve shows that a simple Arrhenius law is followed from which an activation energy of 0.2 eV can be extracted. According to both, high ρ value and thermally activated behavior, it turns out that in Ca₃FeRhO₆, the charge carriers are localized. Such a result is consistent with the antiferromagnetic intrachain coupling. These data for Ca₃FeRhO₆ confirm that despite the existing similarities to isostructural Ca₃CoRhO₆, i.e., the A'_{TP} and B_{oct} crystallographic sites are also occupied by trivalent cations with high spin (S=5/2 for Fe³⁺) and low spin (S=0 for Rh³⁺), respectively, the nature of the magnetic interactions differs strongly.

C. Band structure calculations

For the LSDA band structure calculations we used the augmented spherical wave (ASW) method in its scalar-relativistic implementation [24, 25]. In the ASW method, the wave function is expanded in atom-centered augmented spherical waves, which are Hankel functions

TABLE II: Comparison of the resistivities at room temperature.

compound	$\mathrm{Ca_3Co_2O_6}$	Ca_3CoRhO_6	Ca_3FeRhO_6
$\rho(300\mathrm{K})[\Omega\cdot\mathrm{cm}]$	50 [23]	39 [23]	8 300

and numerical solutions of Schrödinger's equation, respectively, outside and inside the so-called augmentation spheres. In order to optimize the basis set, additional augmented spherical waves were placed at carefully selected interstitial sites. The choice of these sites as well as the augmentation radii were automatically determined using the sphere-geometry optimization algorithm [26]. The Brillouin zone integrations were performed using the linear tetrahedron method with up to 85 k-points within the irreducible wedge. In contrast to our previous work [22] we here use a new version of the ASW code, which takes the non-spherical contributions to the charge density inside the atomic spheres into account.

All calculations are based on the powder data of Niitaka et al. [14]. In a first step, we performed a set of calculations, where spin-degeneracy was enforced. The resulting partial densities of states (DOS) are shown in Fig. 3. While O 2p dominated bands are located in the

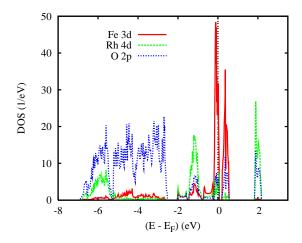


FIG. 3: (Color online) Partial densities of states (DOS) of spin degenerate Ca_3FeRhO_6 .

interval from -6.8 to $-2.4\,\mathrm{eV}$, three groups of bands of mainly metal d-character are found at higher energies. However, the strong d-p hybridization causes large p/d contributions above/below $-2\,\mathrm{eV}$, reaching up to 50% especially for the Rh 4d states.

According to the partial Fe 3d densities of states shown in Fig. 4 the trigonal crystal field at the iron sites results in a splitting into non-degenerate $d_{3z^2-r^2}$ as well as doubly degenerate d_{xy,x^2-y^2} and $d_{xz,yz}$ states. The Rh 4d states as given in Fig. 5 experience a nearly perfect separation of the 4d states into occupied t_{2g} and empty e_g states due to the octahedral crystal field at these sites. Whereas strong σ -type d-p bonding places the Rh 4d e_g states at $2.0\,\mathrm{eV}$, the peak at about $0.4\,\mathrm{eV}$ traces back to Fe $d_{xz,yz}$ states. For this reason, spin-polarization of the latter bands is highly favourable, with the observed high-spin/low-spin scenario.

In a second step, spin-polarized calculations were performed leading to the observed antiferromagnetic ordering, which is by 1 mRyd per Fe atom more stable than

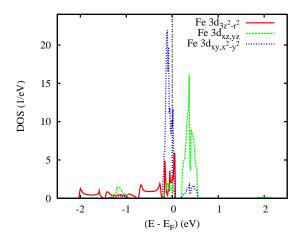


FIG. 4: (Color online) Partial Fe d DOS of spin degenerate Ca_3FeRhO_6 .

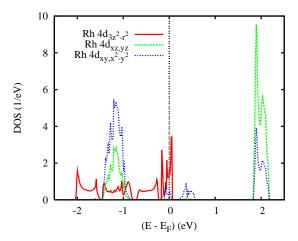


FIG. 5: (Color online) Partial Rhd DOS of spin degenerate $\mathrm{Ca_3FeRhO_6}.$

the ferromagnetic configuration. Well localized magnetic moments of $0.00 \,\mu_B$ (Rh), $3.77 \,\mu_B$ (Fe), $0.13 \,\mu_B$ (O), and $0.01 \,\mu_B$ (Ca) are obtained in close agreement with those of previous calculations [27]. These values reflect the experimental result of low- and high-spin states at the octahedral and trigonal prismatic sites, respectively. The total moment per sublattice amounts to $\pm 4.58 \,\mu_B$, which still might be slightly altered by the inclusion of spin-orbit coupling, which is beyond the present work. In particular, the obtained total magnetic moment per sublattice is smaller than the experimental value deduced from Fig. 1 as was also observed for Ca₃Co₂O₆ [12, 13].

Worth mentioning are the rather high magnetic moments at the oxygen sites arising from the strong d–p hybridization, which sum up to about $0.8\,\mu_B$ per trigonal prism. Adding to the 3d moment they lead to the for-

mation of extended localized moments already observed in $\mathrm{Ca_3Co_2O_6}$ [12] and confirm the formal Fe S=5/2 configuration, hence, the formal Fe³⁺ state. The highspin behavior at the iron sites is clearly observed in the partial DOS shown in Fig. 6, where the Fe 3d minority

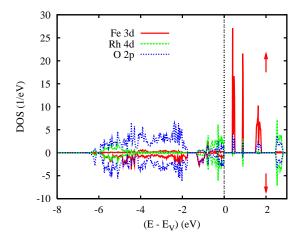


FIG. 6: (Color online) Partial DOS of antiferromagnetic Ca_3FeRhO_6 .

states display sharp peaks above E_F and the spin majority states are spread over a large energy interval as a result of the strong p-d hybridization.

The antiferromagnetic order growing out of the spinpolarized calculations goes along with the opening of an insulating gap of about 0.4 eV as revealed by Fig. 6. This value corresponds to an activation energy of 0.2 eV, which is in remarkably good agreement with the experimental value deduced from Fig. 2.

IV. SUMMARY

In summary, we have performed susceptibility, Mössbauer spectroscopy, and transport measurements on the antiferromagnetic insulating compound Ca₃FeRhO₆. The experimental data have been compared with LSDA band structure calculations, and the agreement is found to be very good. In particular, the calculations confirm several experimental key results as, e.g., the charge and spin states at the Fe and Rh sites including the characteristic high-spin/low-spin scenario, the antiferromagnetic ordering, and the activation energy.

V. ACKNOWLEDGEMENTS

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- [2] For a review, see Colossal Magnetoresistance, Charge Ordering and Related Properties of Manganese Oxides, edited by C. N. R. Rao and B. Raveau (World Scientific, Singapore, 1998) and Colossal Magnetoresistance Oxides, edited by Y. Tokura (Gordon & Breach, London, 1999).
- [3] K. E. Stitzer, J. Darriet, and H.-C. zur Loye, Curr. Opin. Solid State Mater. Sci. 5, 535, (2001).
- [4] H. Fjellvåg, E. Gulbrandsen, S. Aasland, A. Olsen, and B. Hauback, J. Solid State Chem. 124, 190 (1996).
- [5] S. Aasland, H. Fjellvåg, and B. Hauback, Solid State Comm. 101, 187 (1997).
- [6] M. Mekata, J. Phys. Soc. Jpn. 42, 76 (1977); M. Mekata and K. Adachi, J. Phys. Soc. Jpn, 44, 806 (1978).
- [7] H. Kageyama, K. Yoshimura, K. Kosuge, H. Mitamura, and T. Goto, J. Phys. Soc. Jpn. 66, 1607 (1997).
- [8] H. Kageyama, K. Yoshimura, K. Kosuge, M. Azuma, M. Takano, H. Mitamura, and T. Goto, J. Phys. Soc. Jpn. 66, 3996 (1997).
- [9] A. Maignan, C. Michel, A.C. Masset, C. Martin, and B. Raveau, Eur. Phys. J. B 15, 657 (2000).
- [10] A. Maignan, V. Hardy, S. Hébert, M. Drillon, M.R. Lees, O. Petrenko, D. McK. Paul, and D. Khomskii, J. Mater. Chem. 14, 1231 (2004).
- [11] R. Frésard, C. Laschinger, T. Kopp, and V. Eyert, Phys. Rev B. 69, 140405(R) (2004).
- [12] V. Eyert, C. Laschinger, T. Kopp, and R. Frésard, Chem. Phys. Lett. 385, 249 (2004).
- [13] H. Wu, M.W. Haverkort, Z. Hu, D. Khomskii, and L. H. Tjeng, Phys. Rev. Lett. 95, 186401 (2005).
- [14] S. Niitaka, H. Kageyama, M. Kato, K. Yoshimura and K. Kosuge, J. of Solid State Chem. 146, 137, (1999).

- [15] M.J. Davis, M.D. Smith and H.C. zur Loye, J. of Solid State Chem. 173, 122, (2003).
- [16] S. Niitaka, K. Yoshimura, K. Kosuge, K. Mibu, H. Mitamura, T. Goto, J. of Magn. and Magn. Mater. 260, 48, (2003).
- [17] S. Niitaka, K. Yoshimura, K. Kosuge, M. Nishi, and K. Kakurai, Phys. Rev. Lett. 87, 177202 (2001).
- [18] S. Niitaka, K. Yoshimura, K. Kosuge, A. Mitsuda, H. Mitamura, and T. Goto, J. Phys. Chem. Solids 63, 999 (2002).
- [19] V. Hardy, M.R. Lees, A. Maignan, S. Hébert, D. Flahaut, and D. McK. Paul, J. Phys. Condens. Matt. 15, 5737 (2003).
- [20] S. Niitaka, H. Kageyama, K. Yoshimura, K. Kosuge, S. Kawano, N. Aso, A. Mitsuda, H. Mitamura, and T. Goto, J. Phys. Soc. Jpn. 70, 1222 (2001).
- [21] M. Loewenhaupt, W. Schäfer, A. Niazi, and E.V. Sampathkumaran, Europhys. Lett. 63, 374 (2003).
- [22] V. Eyert, U. Schwingenschlögl, C. Hackenberger, T. Kopp, R. Frésard, and U. Eckern, Prog. Solid State Chem., at press (arXiv:cond-mat/0509374).
- [23] A. Maignan, S. Hébert, C. Martin and D. Flahaut, Mater. Sc. and Eng. B 104, 121 (2003).
- [24] A. R. Williams, J. Kübler, and C. D. Gelatt, Jr., Phys. Rev. B 19, 6094 (1979).
- [25] V. Eyert, Int. J. Quantum Chem. 77, 1007 (2000).
- [26] V. Eyert and K.-H. Höck, Phys. Rev. B 57, 12727 (1998).
- [27] A. Villesuzanne and M.-H. Whangbo, Inorg. Chem. 44, 6339 (2005).